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(54) Title: 5,10-METHYLENE-TETRAHYDROFOLATE AS A MODULATOR OF A CHEMOTHERAPEUTIC AGENT			
(57) Abstract The present invention relates to the compound 5,10-methylene-tetrahydrofolate (CH_2FH_4), and its solution isomer FH_4 , therapeutic uses of these compounds, and compositions thereof. CH_2FH_4 and FH_4 strongly modulate the <i>in vivo</i> antitumor effects of 5-Fluorouracil.			

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5,10-METHYLENE-TETRAHYDROFOLATE AS A MODULATOR
OF A CHEMOTHERAPEUTIC AGENT

BACKGROUND OF THE INVENTION

Technical Field

5 The subject matter of the present invention relates to 5,10-methylene-tetrahydrofolate (CH_2FH_4), therapeutic uses of this compound and compositions thereof. CH_2FH_4 strongly modulates the in vivo antitumor effects of 5-Fluorouracil.
10 Furthermore, the present invention additionally relates to a solution isomer of CH_2FH_4 , tetrahydrofolate (FH_4), which also strongly modulates the in vivo antitumor effects of 5-Fluorouracil.

Background Information

15 The compound 5-Fluorouracil (5-FU) is possibly the most widely used anticancer drug in the world. In the 1970s and early 1980s, the prevailing opinion among cancer researchers was that the key biochemical lesion caused by 5-FU in tumor cells 20 resulted from the drug's incorporation into RNA (Kufe et al., J. Biol. Chem. **256**:9802 (1981) and Glazer et al., Mol. Pharmacol. **21**:468 (1982)).

25 In 1982, using a specifically designed assay of the DNA enzyme, thymidylate synthase (TS) (EC 2.1.1.45), the present inventors established that the therapeutic mechanism of 5-FU against murine colon cancer was complete inhibition of TS or abrogation of TS activity (Spears et al., Cancer Res. **42**:450-56 (1982)). In fact, the present 30 inventors were the first to report a clinical correlation between TS level in a patient's cancer after 5-FU treatment and response (Spears et al., Cancer Res. **44**:4144-50 (1984)). The finding has been confirmed by several research groups.

TS is the only intracellular source of new ("de novo") thymine synthesis, as the enzyme which catalyzes the methylation of deoxyuridylate to form thymidylate (thymine-2'-deoxyribose-5'-phosphate).
5 Thymine is one of the four main building blocks of DNA, and its occurrence in DNA (vs. its absence in RNA) is the major structural difference between DNA and RNA. Thus, the activity of TS to make new thymidylate and DNA is essential to cell division,
10 tissue regeneration and turnover, and tumor growth. The source of the methyl one-carbon group for synthesis of thymidylate is CH₂FH₂ and its polyglutamates. The mechanism of methyl transfer by TS has recently been reviewed (K.T. Douglas,
15 Medicinal Res. Rev. 7:441-75 (1987)). After initial weak binding of deoxyuridylate to TS, the enzyme catalyzes ring-opening of CH₂FH₂ at the imidazole C11 ring. This may be the rate limiting step overall. The relative stability of tetrahydrofolate within
20 the ternary complex, toward oxidation, suggests that the ring-opening occurs with the substitution at N5, in accordance with formation of an N5-iminium cation species (S.J. Benkovic, Ann. Rev. Biochem., 49:227-51 (1980)). Covalent bonding between the methylene group and the C5-position of deoxyuridylate is accompanied by rapid hydride transfer from the C6-position of the ring-opened CH₂FH₂ so that CH₃- is formed on the C6 position of the nucleotide. This leads rapidly to expulsion of the two products from
25 the TS binding site(s), i.e., thymidylate and dihydrofolate. TS is the only enzyme which oxidizes reduced folates to dihydrofolate, which is then converted back to tetrahydrofolate by another enzyme, dihydrofolate reductase. In general, the limiting intracellular factors in this biochemical pathway for making thymine are, in order of increasing scarcity, deoxyuridylate, dihydrofolate
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reductase, TS, and then CH₂FH.. Thus, a decrease in thymidine production through the TS pathway can result from nutritional deficiencies which decrease CH₂FH. production (i.e., primary folate deficiency,

5 B12, B6, and other B-vitamin deficiencies which impair folate one-carbon metabolism), or from antimetabolites drugs such as 5-FU or methotrexate. Methotrexate inhibits dihydrofolate reductase, thus blocking the regeneration of tetrahydrofolates from 10 dihydrofolate. 5-FU and other fluorinated pyrimidines (for example, floxuridine, FUDR or trifluoromethylthymidine) block TS activity through formation of the specific metabolite for this 15 effect, fluorodeoxyuridylate (FdUMP), discussed below.

Inhibition of TS activity leads to "thymineless cell death" or "unbalanced cell growth," whereby RNA and protein synthesis, and cell enlargement, occur in the absence of adequate new 20 DNA synthesis (see Goulian et al., Adv. Exp. Med. Biol. 195:89-95 (1986), and refs. therein). In blood cells, such unbalanced cell growth can lead to megaloblastic anemia, macrocytosis, and bone marrow failure.

25 The mechanism of inhibition of TS by FdUMP has been studied intensively for the past two decades (see Santi et al., Biochem., pp. 8606-13, (1987) and refs. therein). In the absence of CH₂FH., FdUMP binds TS extremely weakly. However, in the 30 presence of a large excess of CH₂FH., even low levels of FdUMP will bind tightly to TS, by forming inhibitory TS-FdUMP-CH₂FH. ternary complexes. In the presence of excess CH₂FH., such ternary complexes are stable and no significant TS activity occurs. The 35 molecular basis for the ternary complex is that after CH₂FH. ring-opening to form a covalent bond to FdUMP in the TS enzyme pocket (analogous to the

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normal reaction with deoxyuridylate), no hydride ion transfer can occur. Thus, no dihydrofolate is formed and the covalently-bonded FdUMP-CH₂FH₄ only leaves the enzyme site with great difficulty, as long as free CH₂FH₄ is present in substantial excess. If the CH₂FH₄ concentration is relatively low, the ternary complex dissociates back to starting products, including free, active TS.

Thus, TS inhibition can occur with only trace amounts of FdUMP in slight excess over TS molecules; however, a specific condition must occur in that 5-10-methylenetetrahydrofolate (CH₂FH₄) (and its polyglutamates) must be present in high concentration. Stated more simply, CH₂FH₄ is like a "glue" that holds the FdUMP onto the TS enzyme and therefore inhibits TS activity. However, CH₂FH₄ is also a powerful growth factor, for promotion of purine, protein, and lipid metabolism, as well as pyrimidine synthesis; thus, CH₂FH₄ administration for the purpose of promotion of TS inhibition by FdUMP may be expected to also increase the degree of "unbalanced cell growth."

CH₂FH₄ is a normal intracellular metabolite of the B-vitamin, folic acid, for use in thymidylate synthesis by TS. The same is true with respect to the polyglutamates of CH₂FH₄. However, CH₂FH₄ is also used by several other enzymes including CH₂FH₄ reductase (EC 1.1.99.15), serine hydroxymethylase (EC 2.1.2.1), and C1-tetrahydrofolate synthase and CH₂FH₄ dehydrogenase (EC 1.5.1.5). These interconversions using CH₂FH₄ are essential for purine synthesis, amino acid synthesis (i.e., serine and methionine), and lipid metabolism through the re-methylation of methionine. Thus, CH₂FH₄ is located at a metabolic branch point as a substrate for at least 4 different enzymes (Green et al., Biochem. 27:8014-22, (1988), S.J. Benkovic, Ann.

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Rev. Biochem. 49:227-51 (1980) and Schirch et al.,
Arch. Biochem. Biophys. 269:317-80 (1989)). This
explains the fact that intracellular CH₂FH₂ is
normally present in low concentrations, below 1.0
micromolar. Recent measurements have shown that
intracellular CH₂FH₂ levels are typically low, and
virtually always lower than tetrahydrofolate, using
the bacterial L. Casei TS-[3H]FdUMP ligand binding
assay (Priest et al., Cancer Res. 48:3398-3404
10 (1988), and refs. therein). The present inventors
have modified this assay (Adv. Exp. Med. Biol.
244:98-104 (1988) and Invest. New Drugs 7:27-36
(1989)) and reported relatively low levels of CH₂FH₂,
(much below 1.0 micromolar) in patients' cancer
15 biopsy specimens despite administration of high
doses of leucovorin (LV) (Proc. Am. Soc. Clin.
Oncol. 8:69 (1989)); furthermore, these observations
of the present inventors led to administration of
the amino acid, L-serine, to patients in an attempt
20 to convert the tetrahydrofolates (in various
polyglutamate forms, present in large excess) to
CH₂FH₂ (and polyglutamates). These results have
suggested that increased FH₂, rather than CH₂FH₂, may
be therapeutic. The inventors have recently
25 published the only comparative data that exist for
the different major intracellular one-carbon forms
of folates (Biochem. Pharmacol. 38: 2985-93 (1989)),
showing that of all of these, CH₂FH₂ (at least, as
the monoglutamate) is the best folate form for
30 formation of TS-FdUMP-folate ternary complexes, and
that a concentration of CH₂FH₂ in excess of 1.0
micromolar is desirable for this effect. CH₂FH₂ was
found to be four times stronger than the next best
folate, tetrahydrofolate, and about 100 times
35 stronger than LV.

Leucovorin (referred to as LV, or folinic
acid) is (6R,S)-5-formyl-tetrahydrofolate and has

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been available commercially for decades for the treatment of folic acid (the B-vitamin) deficiency states (The Pharmacologic Basis of Therapeutics, 4th ed. (Goodman et al., eds.) The MacMillan Co., 5 Toronto, pp. 1431-44 (1970)). In 1982, the first clinical reports of the usefulness of LV as a modulator of 5-FU in cancer treatment appeared. (Machover et al., Cancer Treat. Rep. 66:1803-07 (1982)). LV addition to 5-FU appeared to 10 approximately double response rates in patients with gastrointestinal cancers. This result was confirmed in several subsequent studies. (For an extensive review, see Grem et al., Cancer Treat. Rep. 71:1249-64 (1987)). Currently, LV addition to 5-FU therapy 15 is community standard practice in the United States.

The mechanism of leucovorin (LV or folinic acid) improvement in the antitumor therapy of 5-FU and floxuridine (FUDR) has been shown in several studies to be due to improved TS inhibition 20 associated with increased intracellular (6R)-CH₂FH₂ and (6S)-tetrahydrofolates. However, LV appears to be only partially effective in the goal of promoting complete TS inhibition by FdUMP in vivo. For an in vitro example, researchers have shown that TS 25 inhibition after 5-FU, while improved by LV, was still clearly incomplete (Keyomarsi et al., J. Biol. Chem. 263:14402-09 (1988)). In part, this may have been related to saturation of obtainable summed pools of CH₂FH₂ + tetrahydrofolate at about a 5-fold 30 increase over baseline at 30 hr LV exposure. Thus, maximum synergy of LV was obtained at less than 1.0 micromolar exposure, with no further improvement at higher concentrations although human plasma folates (LV and methyltetrahydrofolate, MTHF) are higher 35 than this after high-dose LV administration (Doroshow et al., NCI Monogr. 5:171-74 (1987)). A related observation may be that addition of high-

dose folic acid (140 mg/m^2) to 5-FU therapy appears to be associated with an increase in toxicity without improved response rates (Asbury et al., Am. J. Clin. Oncol. **10**:47-49 (1987)).

5 In fact, decreasing synergy has been shown for LV addition to FUDR at concentrations above 0.5 micromolar, when the colon cancer cells were previously folate-deficient (Davis et al., Mol. Pharmacol. **35**:422-27 (1989)). Also, others have
10 shown in vivo in mice that expansion of breast tumor CH₂FH₂ pools was a maximum of less than two-fold over baseline despite massive LV dosing (180 mg/kg x 8 over 48 hr) (Wright et al., Cancer Res. **49**:2592-96 (1989)). These observations are mirrored in recent
15 clinical trials comparing the therapeutic outcome in colon cancer, in which low-dose LV (20 mg per square meter) was more effective than high-dose LV (200 mg per square meter) in terms of both tumor response rate and patient survival (Poon et al., J. Clin. Oncol. **7**:1407-18 (1989)). The lack of effectiveness of high-dose LV in promoting complete TS inhibition was suggested by researchers based on tumor biopsy analyses in breast cancer patients: LV increased TS inhibition from an average of 30 ± 13 to 71 ± 14 %,
20 with responding patients showing the higher percentages of TS inhibition than non-responders (Swain et al., J. Clin. Oncol. **7**:890-99 (1989)).

In view of the above, the present inventors realized the potential of the direct administration of CH₂FH₂ to patients receiving 5-FU, as such a course of action would maximize TS inhibition.

The desirability and ability to use CH₂FH₂ in the method of the present invention have never been obvious for various reasons.

For example, CH₂FH₂ as a compound in solution has enjoyed a general reputation of being

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extremely unstable. (Temple et al., "Chemical and Physical Properties of Folic Acid and Reduced Derivatives," In Folates and Pterins (Blakely et al., eds.), Vol. 1, pp. 61-63 (1984) and Wright et al., Cancer Res. 49:2592-96 (1989)). In solution, it is generally known to exist in equilibrium with FH₄, requiring excess formaldehyde to favor the equilibrium toward CH₂FH₄.

10 Under anaerobic conditions, such as made possible for clinical administration of CH₂FH₄ by a closed, delivery system (U.S.Patent 4,564,054), powdered tetrahydrofolate is stable even at room temperature, for a year or more (Caldwell et al., Prep. Biochem. 3:323-26 (1973)).

15 Additionally, published data on the clinical tissue levels of CH₂FH₄ in patients have been limited, and it is well known that LV can be given in gram-size doses (Grem, et al., supra). LV is an extremely powerful folate (B-vitamin) that is 20 one-hundred times stronger than folic acid in correcting nutritional folate deficiency. As little as 1.0 mg of LV will correct folate deficiency as a single dose (The Pharmacological Basis of Therapeutics, supra). Thus, it is logical to 25 assume that tumor CH₂FH₄ levels might reach saturation levels from high dose LV.

Finally, it appears that no published studies exist on the toxicological aspects of CH₂FH₄. More specifically, there seems to be no 30 available published work on either in vitro or in vivo effects of direct exposure of living cells to CH₂FH₄.

Thus, in view of the structural properties 35 of CH₂FH₄, as well as the lack of information regarding the effects of CH₂FH₄, the present invention is quite remarkable. CH₂FH₄ is utilized to

potentiate or modulate the antitumor effects of the chemotherapeutic agent 5-FU.

L.R. Hughes (Eur. Pat. Appl. EP 284,3380 and Chem. Abstr. 110:95789 (1989)) has described a 5 novel folate analog as a TS inhibitor and antitumor agent. However, the discovery is clearly radically different from the present invention. The analog does not occur naturally, is absent two nitrogen atoms, is not reduced, and has a reactive propargyl 10 group attached to the glutamate moiety. Also, no mention is made of 5-FU.

Interleukin-2 has been proposed as a modulator of tetrahydrobiopterin (US Patent 4,752,573); however, interleukin-2 is an 15 oligopeptide having no resemblance to leucovorin, and no claim for TS inhibition or interaction with 5-FU is made.

A patent for radiolabeled assay of folates (US Patent 4,136,159) has no therapeutic 20 pharmaceutical intent, and makes no mention of TS inhibition.

Various patents exist for other, unnatural folate analogs, including quinazolines and dideazatetrahydrofolates as inhibitors of enzymes 25 such as folylpolyglutamyl synthetase (e.g., see Chem. Abstr. 110: P39366p (1989)). However, these are unnatural analogs which have distinct chemical, structural differences from CH₂FH₂.

The European patent application (EP 30 266,042) of Wood et al. describes a process for separation of diasteriomers of LV, as well as (6R)- and (6S)-tetrahydrofolates. No use of CH₂FH₂ as a 35 potentiator of TS inhibition by FdUMP (and thus 5-FU and other fluoropyrimidines) is claimed in the document.

All U.S. patents and publications referred to herein are hereby incorporated by reference.

SUMMARY OF THE INVENTION

The present invention relates to the compound CH₂FH₄ and its solution isomer FH₂, therapeutic uses of these compounds, and 5 compositions thereof. CH₂FH₄ and FH₂ strongly potentiate the antitumor or TS-inhibitory effects of 5-FU.

More specifically, the present invention includes a method of inhibiting the growth of a tumor in a patient comprising administering to said patient an amount of parent CH₂FH₄ or FH₂, and 5-FU sufficient to effect said growth inhibition. The CH₂FH₄ or FH₂, may be administered concurrently with 5-FU, or prior to the administration of 5-FU. In the latter case, the CH₂FH₄ or FH₂, is administered 6-24 hours, or preferably 1-3 hours, before the administration of the 5-FU.

The CH₂FH₄ or FH₂, may also be administered after the administration of 5-FU in which case the CH₂FH₄ or FH₂, compound is administered 1-10 days, or preferably 1-6 hours, after the 5-FU administration.

Furthermore, the CH₂FH₄ or FH₂, solution may be administered either intravenously, intraarterially, or intraperitoneally, and in a dosage of 5-500 mg/m² (body surface area). Preferably, it may be administered in a dosage of 20-200 mg/m² (body surface area). The CH₂FH₄ or FH₂, solution may also be administered orally or topically as a 0.5% cream under an occlusive dressing.

If it is administered intravenously, such as through a central venous catheter, the CH₂FH₄ or FH₂, solution may be given in a dosage of 5-500 mg/m² (body surface area), or preferably 20-200 mg/m², every 4-6 hours, once daily, or once weekly or as a

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continuous infusion of 20-200 mg/m²/week.

Additionally, if it is administered every 4-6 hours, the CH₂FH₂ or FH₂ solution may be administered prior to, or subsequent to, the administration of

5 5-FU.

The CH₂FH₂ or FH₂ may be administered as the 6R, 6S, or as a mixture of the 6R and 6S enantiomers (diastereomers).

Also, if the CH₂FH₂ or FH₂ is administered 10 in an alkaline vehicle, the concentration of the CH₂FH₂ or FH₂ is from 0.1 to 20 mg/ml whereas if the compound is administered in physiologic saline, the concentration is from 0.1 to 10 mg/ml.

Furthermore, the present invention 15 includes a method of using CH₂FH₂ or FH₂ in order reduce the toxicity of an anti-folate drug which has been administered to a patient. Examples of anti-folate drugs include methotrexate, trimetrexate, nitrous oxide, and dideoxytetrahydrofolic acid.

The present invention also includes a 20 method of treating folate deficiency states by the administration of CH₂FH₂ or FH₂.

Moreover, the present invention also 25 includes a method of treating B12- and B6-refractory anemias whereby CH₂FH₂ or FH₂ is administered in an amount sufficient to effect said treatment.

Furthermore, the present invention also 30 includes a composition containing CH₂FH₂ or FH₂ and 5-FU, as well as a pharmaceutically active carrier. The composition may also contain a stabilizing agent such as an ascorbate salt, or glutathione. The composition may also contain free formaldehyde.

Additionally, the present invention also 35 includes a composition containing CH₂FH₂ or FH₂ and a compound which is metabolized to FdUMP, as well as a pharmaceutically active carrier. Examples of

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compounds which can be metabolized to FdUMP include floxuridine (FUDR), fтораfur (tegafur), and 5'-deoxyfluorouridine (Doxifluridine®).

The composition may also contain a stabilizing

5 agent, such as an ascorbate salt, or glutathione.

Formaldehyde may also be present in the composition.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 represents the effect of CH₂FH₄ ("CH₂H,PteGlu,_n") on TS inhibition in 5-FU-resistant 10 colon cancer cells (from tumor 51) after the administration of 5-FU ("FUra").

Figure 2 represents the structure of (6R,S)-methylene-tetrahydrofolic acid (or CH₂FH₄) and the configuration of the natural (6R)-CH₂FH₄ 15 enantiomer (diastereomer) (Poe et al., Biochem. 18:5528 (1979) and Kalbermatten et al., Helv. Chim. Acta 64:2633 (1981)).

Figure 3 represents the structure of tetrahydrofolic acid or FH₄, the predominant form at 20 concentrations of less than 1 mM.

Figure 4 shows the results of TS-[³H]FdUMP-folate binding assay of CH₂FH₄ as a function of concentration of the folate in 0.2 M Tris buffer, pH 7.4, with and without formaldehyde (CH₂O), 6 mM, 25 addition.

DETAILED DESCRIPTION OF THE INVENTION

One embodiment of the present invention relates to the use of CH₂FH₄ as a modulator of 5-FU 30 in cancer chemotherapy. CH₂FH₄, as well as FH₄, increase response rates to 5-FU as a result of increasing the inhibition of TS by the 5-FU metabolite, FdUMP, in tumors. Thus, CH₂FH₄ can be used to inhibit the growth of tumors when used in combination with 5-FU, or with other drugs which are

metabolized to FdUMP including floxuridine (FUDR), florafur (tegafur), and Doxifluridine® (5'-deoxyfluorouridine).

The mechanism of action of CH₂FH₄ is promotion of TS inhibition by FdUMP in fluoropyrimidine-treated tumors, which can occur by increasing the rate of formation and stability of TS-FdUMP-CH₂FH₄ and TS-FdUMP FH₄ ternary complexes. Administration of CH₂FH₄ in doses ranging from 5-500 mg/m² (body surface area), or preferably 20-200 mg/m², will result in expansion of intracellular pools of both CH₂FH₄ and FH₄ as monoglutamates. These are the best two folate forms as substrates for polyglutamation, the major intracellular forms for retention of folates, as well as for direct binding to TS-FdUMP complexes. One carbon exchange between endogenous CH₂FH₄-polyglutamates and tetrahydofolate-monoglutamate resulting from CH₂FH₄ administration, as suggested in Tables II and III, would indicate that the optimal times for bolus 5-FU administration are concurrently or at several hours after bolus I.V. CH₂FH₄ administration and thus after maximum polyglutamation. CH₂FH₄ may also be administered after 5-FU is given or as a protracted, continuous infusion.

More specifically, CH₂FH₄ may be administered 6-24 hours, or preferably, 1-3 hours, prior to the administration of 5-FU. CH₂FH₄ can also be administered 1-10 days, or preferably 1-6 hours, subsequent to the administration of 5-FU.

Polyglutamation of folates causes retention within the cell, and typically also accelerates rates of enzyme processing of one-carbon interconversions of folates (Schirch et al., Arch. Biochem. Biophys. 269:371-80 (1989), Green et al., Biochem. 27:8014-22, 1988). Current data would suggest that polyglutamation of FH₄ and CH₂FH₄ will

promote TS-FdUMP-folate inhibitory ternary complex formation to a greater extent than promotion of the normal enzymic reaction with deoxyuridylate

(Houghton et al., Cancer Res. 48:3062-69 (1988)).

5 Since polyglutamates may form TS-FdUMP-folate ternary complexes as much as 50-fold more tightly than parent monoglutamates, an objective of folate addition to fluoropyrimidine therapy could also include formation of TS-FdUMP-tetrahydrofolates, 10 which would also be strongly inhibitory. In addition, a role for the unnatural enantiomers (diastereomers at the pterin C6- position), such as polyglutamates of (6S)-CH₂FH, or (6R)- 15 tetrahydrofolate, in TS inhibition by forming TS-deoxyuridylate-folate or TS-FdUMP-folate ternary complexes, potentially could be a factor (Kisliuk et al., Biochem. 20:929-34 (1981)) in the TS inhibition observed with CH₂FH administration in vivo (Tables I, II, and III; Fig. 1).

20 The potentiation of TS inhibition by low levels of FdUMP may be expected to last only a few hours unless polyglutamation of the CH₂FH, and FH, occurs thereby creating more powerful TS-FdUMP binders than the parent monoglutamate. Thus, CH₂FH, 25 dosing requirements may be as frequent as every 4-6 hrs., once daily, or as infrequent as once weekly.

In one embodiment of the present invention, CH₂FH, can be administered by intermittent (e.g., daily) bolus dosing in patients who have 30 central venous catheters. Such patients could self-administer the CH₂FH, (using a means for ensuring the stability of the formulation to oxidation) and would also be candidates for administration of CH₂FH, by continuous, intravenous protracted infusion. The 5- 35 FU infusion would be expected to produce low levels of FdUMP in tumors. Low FdUMP levels would be expected to be associated with relatively poor TS

inhibition unless CH₂FH₂ levels were very high. FH₂, free of formaldehyde as a stabilizer may also be administered in the same manner.

An ameliorating factor to consider may be that chronic TS inhibition, albeit incomplete, would be expected to cause slight increases in CH₂FH₂ levels because of lowered consumption of CH₂FH₂ in the natural TS mechanism so that pharmaceutical CH₂FH₂ in this setting might be more efficient.

Other embodiments include the addition of CH₂FH₂ at late times after bolus intravenous 5-FU infusion (e.g., at 6 hours in the daily 25 (monthly) Schedule, or at days 4, 5 and 6 on the biweekly bolus schedule.)

In addition to being administered intravenously, CH₂FH₂ may also be administered intraarterially or intraperitoneally, also in a dosage of 5-500 mg/m², or preferably, in a dosage of 20-200 mg/m². However, CH₂FH₂ may also be administered topically as a 0.5% cream under an occlusive dressing.

Another embodiment of the present invention comprises a composition containing CH₂FH₂ as well as 5-FU. The composition also contains a pharmaceutically active carrier, and may also contain formaldehyde in excess as a stabilizer.

A further embodiment of the present invention includes a composition containing CH₂FH₂ and one or more other drugs which can be metabolized to FdUMP. The composition may contain a pharmaceutically active carrier, and may also contain formaldehyde in excess as a stabilizer.

It should be noted that FH₂, free of formaldehyde, can replace the use of CH₂FH₂ in each of the above embodiments.

Because reduced folates are rapidly interconvertible according to their one-carbon

states, it may be anticipated that the clinical tolerance for CH₂FH₄ or FH₄ will be similar to that of LV and 5-methyl-tetrahydrofolate (MTHF), the latter of which is the predominant blood transport form of folates.

Also, tetrahydrofolate, and possibly CH₂FH₄, have recently been reported as accumulating to low but significant (i.e., less than 20 micromolar) concentrations in human plasma after LV administration to human subjects (Bunni et al., Cancer Chemother. Pharmacol. 23:353-57 (1989)).

Thus, it can be anticipated that the dose tolerance for CH₂FH₄ or FH₄ in humans is similar to the reported experiences with LV and methyltetrahydrofolate (MTHF) (both of which are given as a mixtures of enantiomers). Specifically, an upper limit of 500 mg per square meter body surface area would be expected to be therapeutically effective. The lowest effective dose may possibly be more powerful than either LV or MTHF, and thus could be as low as 5 mg per square meter body surface area in a single dose. A dosage of 20-200 mg/m² (body surface area) is preferred.

Based on previous studies of the toxicology of folates (LV, MTHF and folic acid) combined with 5-FU and fluorodeoxyuridine, the LD₅₀ in rats would be expected to be above 150 mg/kg i.v. (single bolus) with regard to CH₂FH₄ or FH₄, and may be expected to cause convulsions in such high doses (Bartosek et al., Chimioterapia Onkologica 2(4): 85-98 (Dec. Supp, 1987)).

The pH of the CH₂FH₄/FH₄ solution which is to be injected, may range from slightly acidic to slightly alkaline. 5-FU up to 50 mg/mL in alkaline media may be present, analogous to the practice of formulation of 5-FU and LV in the same solution (e.g., Trave et al., J. Clin. Oncol. 6:1184-91

(1988)). Furthermore, the concentration for injection may be as high as 100 mg/10 mL, preferably from 0.1 to 20 mg/ml, in alkaline vehicles. The concentration may also be as high as 100 mg/20 mL, 5 preferably from .1 to 10 mg/ml, in physiologic, normal saline. At concentrations less than 1 mM in initial CH₂FH₂ concentrations, the predominant form in solution is FH₂ (i.e., the dilution of CH₂FH₂ in aqueous solution shifts the equilibrium between FH₂, 10 and CH₂FH₂, towards FH₂, regardless of pH, O₂ tension, or the presence of reducing agents).

Ascorbate salts may be present as stabilizers (e.g., 1% w/v as the salt at neutral or slightly alkaline pH). Other reducing substances 15 may also be used as stabilizers, for example, reduced glutathione.

Free formaldehyde (CH₂O) may also be present in concentrations up to 10 mM. However, the dosage must be adjusted for formaldehyde toxicity. 20 The formulation may be made directly from (6R,S)-FH₂ powder, alternatively. In this case, formulations would be checked and controlled for the degree of spontaneous condensation of formaldehyde from ambient air to form CH₂FH₂. The oral LD₅₀ (or 25 lowest lethal dose) of CH₂O in humans has been reported to be 36 mg/kg (Registry of Toxic Effects of Chemical Substances, US DHHS, PHS, CDC, NIOSH, Vol. 1, p. 822 (1980)). The pure (6R)CH₂FH₂ or (6S)FH₂ enantiomer may also be utilized, free of the 30 non-TS-binding, unnatural (6S)CH₂FH₂ or (6S)FH₂ enantiomer, respectively. Enantiomer separation is obtainable by chiral column or DEAE column preparative isolation (Kaufman et al., J. Biol. Chem. 238:1498-1500 (1963)).

35 A major advantage of CH₂FH₂ over FH₂ as the parent powdered material is the protection against oxidation, referred to above, which protection would

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therefore be greater with concentrated versus dilute (e.g., < 0.5 mM) concentration, in the absence of a mechanism for excluding air during reconstitution and administration (as provided by the Protector 5 device).

It appears that direct administration of CH₂FH₂ or FH₂, either as the mixture of 6R and 6S diastereomers (enantiomers), the unnatural 6S-CH₂FH₂, or the natural 6R-CH₂FH₂, alone (or their FH₂ solution 10 equilibrium products) can overcome some of the disadvantages of LV described above. That is, CH₂FH₂ addition to 5-FU can lead to greater tetrahydrofolate and CH₂FH₂ elevations intracellularly than LV or MTHF (which both require 15 one carbon activation), and consequently show more profound synergism on TS inhibition by FdUMP.

The applications for CH₂FH₂ or FH₂ are quite significant and far-reaching. For example, antitumor uses of CH₂FH₂ or FH₂, combined with TS- 20 inhibitory fluoropyrimidines include: 1) addition to Platinol/5-FU infusion therapy in head and neck cancer and other epidermoid cancers, 2) addition to combination cyclophosphamide/doxorubicin/5-FU in breast cancer 3) addition to topical Efudex® (5-FU) 25 cream under an air-free occlusive dressing for skin conditions (for example benign keratoses, keratoacanthomas, verrucae, premalignant keratoses, *in situ* cancer and invasive superficial malignancies amenable to topical therapy). Furthermore, CH₂FH₂ or FH₂, can also be applied to those cancer types in 30 which 5-FU and floxuridine are typically combined with LV, such as in colon, rectal and pancreatic carcinomas.

CH₂FH₂ or FH₂ can also be utilized with 35 respect to non-malignancy related conditions. For example, CH₂FH₂ or FH₂, can be used with respect to B12- and B6-refractory anemias which are not

responsive to LV. CH₂FH₂ or FH₂ can also be used to treat folate deficiencies. Furthermore, CH₂FH₂ and FH₂ can also be used for the potentiation (selective rescue of the host patient) of the TS inhibitory mechanism of antibacterial action of nucleotide analogs.

5 Additionally, CH₂FH₂ or FH₂ can be utilized to reduce the toxicity of anti-folate drug which have been administered to patients. Such anti-
10 folate drugs include, for example, methotrexate, trimetrexate, nitrous oxide, and dideoxytetrahydrofolic acid.

15 As a rescue agent following methotrexate, CH₂FH₂ or FH₂ may be more specific than the presently used LV (or MTHF) since CH₂FH₂ would require less (or no) metabolic activation in the case of FH₂ to provide for purine, pyrimidine, and the amino acid synthetic requirements normally met by intracellular folates. CH₂FH₂ could also therefore become useful
20 in rescue of the host in the trimetrexate treatment of Pneumocystis carinii infections of immunosuppressed patients (i.e., AIDS patients).

The present invention can be illustrated by the use of the following non-limiting examples.

25

Example 1Synthesis of CH₂FH₂ as a Low-Formaldehyde MaterialPreparation of (6R,S)-CH₂FH₂:

CH₂FH₂, as the equal mixture of diastereomers (optical isomers or enantiomers at the C6-position; both diastereomers are of the natural L-configuration at the alpha-carbon position of the glutamate moiety) was prepared from (6R,S)-
30 tetrahydrofolic acid, commercially available from Sigma, in the examples described below. The method
35 of synthesis has been described previously (C.P.).

Spears and B. Gustavsson, Adv. Exp. Med. Biol. 244:98-104 (1988)). To (6R,S)-tetrahydrofolate powder, (100 mg) is added 360 μ L of 1.0 M Na Ascorbate, pH 6.5, 68 μ L of 37% (w/w) formaldehyde (CH₂O), and 16 mL phosphate buffer, pH 7.0. A 10-min room temperature incubation allows completion of formation of (6R,S)-CH₂FH₂. This material is applied to a DEAE-cellulose column using a modification of a well-known procedure (Kaufman et al., J. Biol. Chem. 238:1498-1500 (1963)). A step elution with NH₄HCO₃ buffers of increasing concentration and pH, leads to isolation of CH₂FH₂ in the last pooled fraction. This material does not contain free formaldehyde as assayed colorimetrically by toluene extraction of dimedone (methone)-trapped [¹¹C] CH₂FH₂, prepared with [¹⁴C]CH₂O as described previously (Moran et al. Proc. Natl. Acad. Sci. USA 76:1456-60, 1979). Phosphate buffers and TEAE-cellulose can also be used in the procedure of Kaufman, which gives both enantiomers of CH₂FH₂ in the same peak; however, if potassium bicarbonate buffer is used, a separation of the enantiomers is effected, with the biologically active, natural-configuration, (6R)-CH₂FH₂ peak eluting after the (6S)-CH₂FH₂ peak. The amount of formaldehyde (as methylene) in the product may, in fact, be even less than stoichiometric with tetrahydrofolate (Horwitz et al., J. Med. Chem. 12:49-51 (1969)). The amount of (6R)-CH₂FH₂ in the preparations is checked by one or more of the three following methods. (1) Spectrophotometrically, by use of this material as the limiting substrate in a TS assay with L. Casei enzyme, as described by Daron et al. (J. Biol. Chem. 253:940-45 (1978)); (2) ligand binding assay using [⁶⁻³H]FdUMP and L. Casei TS described by the inventors (Adv. Exp. Med. Biol. 244:98-104, 1988); and by absorbance at 294 nm on HPLC (Lu et al., Biochem. 23:6870-75 (1984)).

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Column-isolated CH₂FH₂, whether racemic in 6R- and 6S-forms or as the 6R-form alone in solution can be stored under argon at -80°C for up to a year without decomposition (Bruice, et al. Biochem. 21: 6703-09 (1982)). Alternatively, solutions of CH₂FH₂ after column isolation can be lyophilized to powder and stored under nitrogen in sealed glass ampoules. Various ratios of formaldehyde to CH₂FH₂ can be used, from less than stoichiometric, as described above, including no formaldehyde (either bound as methylene, or free) to a 2- to 4-fold or more excess (Bruice, et al., Biochem. 21:6703-07, (1982)). The use of 2-mercaptoethanol or other reduced thiols has been advocated by some workers, but is unnecessary and may cause minimal interference (S.F. Zakrewski, J.Biol.Chem. 241:2957-961 (1966) and Kallen et al. J.Biol.Chem. 241:5845-50 (1966)) in condensation of CH₂O with tetrahydrofolate.

Alternative methods for synthesis and purification of (6R,S)-CH₂FH₂ are reviewed by C. Temple, Jr. and J.A. Montgomery, In: Folates and Pterins (R.L. Blakley and S.J. Benkovic, eds.), vol. 1, Chemistry and Biochemistry of Folates, John Wiley & Sons, New York, pp.61-120 (1984). This includes use of (6R,S)5-formyltetrahydrofolate (LV), which is commercially available in bulk quantities, and is converted to the 5,10-methenyl-tetrahydrofolate by acidic conditions. The latter compound then can yield CH₂FH₂ by reduction with borohydride in DMSO and pyridine (Farina et al., J. Am. Chem. Soc. 95:5409 (1973)).

Preparation of (6R)-CH₂FH₂:

The naturally-occurring diastereomer (enantiomer) of CH₂FH₂, (6R)-CH₂FH₂, can be prepared by a number of methods, including that of Kaufman et al. as described in the foregoing section, using

TEAE-cellulose elution by bicarbonate.

Commercially-available folic acid reduced to dihydrofolate using hydrosulfite (Mathews et al.

J.Biol.Chem. 235:3304-08, (1960)) or dithionite

5 (R.L. Blakley, Nature 188:231-32, (1960)) is used as a substrate for purified dihydrofolate reductase in the presence of NADPH (e.g., see M. Poe et al, Biochem. 18:5527-30 (1979)). Formation of (6S)-tetrahydrofolate (which is the natural diastereomer)

10 is readily followed at 294 nm. Purification is then done by chromatography (e.g., S.F. Zakrewski and A.M. Sansone, Methods Enzymol. 18B:728-31, 1971), followed by lyophilization to powder and storage under nitrogen or argon in sealed glass vials.

15 An additional approach is reduction of dihydrofolic acid by dihydrofolate reductase in the presence of formaldehyde (Horne et al., Methods Enzymol. 66:545ff (1980)), followed by column isolation, which avoids the need for a separate CH₂O step after (6S)-tetrahydrofolate isolation. In these preparations, ascorbate is typically present (e.g., 0.1M) as an antioxidant. Synthesis of the unnatural (6R)-CH₂FH₂ isomer has been described, by selective enzymic conversion of (6R)-CH₂FH₂ to 20 dihydrofolate, which is easily separated by column chromatography (Anal. Biochem., Vol. 154, pp 516-24 (1986)). The isomeric solution of (6S)-FH₂ is 25 obtained by dilution to less than .5 mM.

Stability of CH₂FH₂:

30 Solutions of CH₂FH₂, as well as the powder, are unstable in the presence of oxygen, with oxygen degradation being catalyzed by light, acid, base, and heavy metals (R.G. Kallen, Methods Enzymol. 183:705ff, 1971). CH₂FH₄ is somewhat more 35 stable than FH₂, as are the major N5-substituted tetrahydrofolates; FH₂ solutions can undergo 90%

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degradation in 4.1 hr when exposed to air (discussed in C. Temple, Jr., and J.A. Montgomery, supra.

However, tetrahydrofolate is completely stable under anaerobic conditions Caldwell et al., Prep. Biochem.

5 3:323-26 (1973).

Thus, a method for air-free reconstruction of CH₂FH₂ or FH₂ powder (in vacuum, or under nitrogen or argon in air-tight ampoules), or fresh handling of column-isolated CH₂FH₂ or FH₂, is required to ensure the stability of CH₂FH₂ as a pharmaceutical with accurate dosing. The invention of Gustavsson, one of the present inventors, (U.S. Patent 4,564,054) referred to as the Protector device, affords such a method. The Protector invention is not generally known, since it is marketed as a method for prevention of aerolization of mutagenic/toxic cancer chemotherapy agents, however, it is equally useful for air-free reconstitution, dosing, and i.v. administration of drug solutions to patients. The Protector is suitable for handling all anticipated dose ranges and concentrations of CH₂FH₂, with the volume for dosing limited only by the syringe size. Vehicles for reconstitution of CH₂FH₂ or FH₂ powder include 5% dextrose, normal (0.89% w/v) saline, 5-FU solutions, and sterile water, (which may or may not be de-aerated for removal of dissolved oxygen prior to use in reconstitution of CH₂FH₂ or FH₂ powder, depending on the presence in the formulation of antioxidant stabilizers such as ascorbate). The Protector may be modified to use semi-opaque materials, such as brown plastic, to reduce transmission of ambient light.

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Example 2CH₂FH₂ USE WITH 5-FU IN MURINE COLON CARCINOMA CA51

(6R,S)-CH₂FH₂ was prepared by the DEAE-cellulose column procedure, described above, using step-elution of the material as previously reported for purification of nucleotides (Moran et al., Proc. Natl. Aca. Sci. USA 76:1456-60 (1979)). To twenty micromoles of (6R,S)-FH₂ (Sigma) were added 62.5 ul of 1.0 M Na Ascorbate, pH 6.5, 2.7 ul of 37% formaldehyde stock, and 0.6 mL of 5 mM phosphate buffer, pH 7.0. Because of the high formaldehyde, this solution was over 2 mM in CH₂FH₂, with less FH₂ present as the solution isomer. After 20 min at room temperature, this solution was applied to a 1 x 3-cm DEAE-cellulose column; in the last step, the 500 mM NH₄HCO₃ (pH 8.0) fraction (30 mL) was pooled, lyophilized to dryness, and stored under vacuum in glass ampoules. Spectrophotometric assay of powder reconstituted in phosphate-buffered-saline showed a concentration of (6R)-CH₂FH₂ in this solution of 2.4 mM; prior assay by L. casei TS-[³H]FdUMP-folate ternary complex formation gave a concentration of 2.5 mM.

On the day of reconstituting the above CH₂FH₂, mice bearing subcutaneous murine colon carcinoma Tumor 51 were administered intraperitoneal (i.p.) 5-FU, with or without concomitant i.p. CH₂FH₂ by separate injection. The 5-FU was given at a dose of 1.6 mg per mouse, about 80 mg/kg. The CH₂FH₂ was given at a dose of 0.5 mL of the 2.4 mM material (1.2 mmole/mouse), above. The in vivo methodologies were essentially as had previously been described (C.P. Spears, et al., Cancer Res. 42:450-56 (1982)). In contrast, however, to the extensive prior experience of the present inventors with this 5-FU-

resistant tumor line, which always had shown significant FdUMP-titratable free TS levels, the tumors of mice receiving concomitant CH₃FH, showed abrogation of TS activity (Table I and Figure 1).

5 The free TS levels of the 5-FU-only treated mice were comparable to the previous observations of the inventors in this line, and at the 1.0 pmol/g level of TS activity was sufficient to support thymidylate synthesis required for tumor growth (C.P. Spears,

10 Exerpta. Med. Int. Congr. Series 647:12-19, (1984)). The levels of apparent free TS in tumors of mice receiving CH₃FH, concomitant with 5-FU were at, or below, that level due to exchange-labeling of endogenous TS-FdUMP-folate ternary complexes in the

15 cytosolic extracts. Stated otherwise, the average \pm S.D. apparent TS value of 0.42 \pm 0.20 pmol/g for the 5 tumors of the 5-FU + CH₃FH, treatment group when corrected downward for labeling of endogenous FdUMP-inhibited enzyme by a minimum correction factor of

20 5% (Spears and Gustavsson, Adv. Exp. Med. Biol. 244:98-104, (1988)) equates with zero detectable TS activity. This is exactly the qualitative difference between sensitivity and resistance to 5-FU previously established (see Spears et al., Cancer Res. 42:450-52 (1982)). An additional observation

25 was that in the Tumor 51 specimens from mice receiving CH₃FH, concomitant with 5-FU was that the pre-incubation dissociation condition, which had previously been routinely used for regenerating all

30 TS in the free form, was completely unable to regenerate free TS, in contrast to the more normal findings in the 5-FU-only exposed tumors. This is strongly suggestive that CH₃FH, administration raised concentrations of tumor CH₃FH, and FH, so high, that

35 even after large dilution into the assays the concentrations were still above those that could

spontaneously oxidize to lower levels permitting in vitro ternary complex dissociation.

The results obtained from Example 2 are shown in Figure 1, and in Table I.

TABLE I

TS INHIBITION IN MURINE TUMOR CA51 AFTER 5-FU^a
EFFECT OF CO-ADMINISTRATION OF CH₂FH.^b

(Values = Ave. ± S.D.)

Hours	<u>5-FU Alone</u>		<u>5-FU + CH₂FH.</u>	
	Free TS ^c (pmol/g)	% Inhibition	Free TS ^c (pmol/g)	% Inhibition
1	1.67 ±0.28	83.3 ±2.8	0.41 ±0.26	95.9 ±2.6
			0.164 ±0.13	98.4 ±1.3
3	1.00 ±0.72	90.0 ±7.2	0.36 ±0.06	96.4
			0.71 ±0.03	92.9
6	1.27 ±0.06	87.3 ±0.6	0.46 ±0.05	95.4

^a 80 mg/kg i/p.

^b 27 mg/kg in (6R) CH₂FH, by spectrophometric and binding assays.

^c Not corrected for ternary complex exchange labeling or ratio of CH₂FH, to FH,. A minimal correction factor of 5% leads to the calculation that there was 100% TS inhibition for all tumors receiving the combination of 5-FU and CH₂FH,, compared to only 92% average TS inhibition by 5-FU alone. Baseline total TS was 10.00 ± 0.04 pmol/g.

Example 3

CH₂FH₂ was formulated, assayed, and administered to 2 patients who had previously been treated with 5-FU. The assays were performed by the methods described in Spears et al., Adv. Exp. Med. Biol. 244:98-104 (1988). In the data shown, the TS inhibition profiles that resulted from CH₂FH₂ administration were not due to concurrent 5-FU dosing. The most recent exposure to 5-FU in these cases was slightly greater than a week prior to the study date, with the patients eligible, however, from the standpoint of toxicity evaluation to receive the weekly dose of 5-FU. Thus, residual FdUMP levels from previous exposure, below the detectable limits for assay, were expected to be present (See Spears et al. Mol. Pharmacol. 27:302-07 (1985)). The serial biopsies were done following single dose administration of CH₂FH₂.

The formulation of CH₂FH₂ was as described in Example 2, and was performed on the day of CH₂FH₂ administration. The assays were also performed on the day of CH₂FH₂ administration.

The results in these patients of the pharmacodynamic tumor tissue analyses showed striking evidence of TS inhibition following CH₂FH₂ administration. These results are summarized in Tables II and III below.

TABLE II

TS INHIBITION AFTER CH₂FH₂ ADMINISTRATION

PATIENT: A.M.; last 5-FU treatment: ≥ 1 week
LOCATION: Östra Sjukhuset (Eastern Hospital), Sweden
TUMOR: Skin metastasis from gastric carcinoma
CH₂FH₄ FORMULATION: 0.1 M Na Ascorbate, pH <9.5, Sigma
(6R,S)CH₂FH₄, DEAE-column purified
CH₂FH₄ DOSE: 30 mg in 30 cc IV over 2 min; 4 mg
as parent CH₂FH₄,
26 mg as FH₄.

(Tumor Tissue Values = Ave. \pm S.D.)

Time of Biopsy ^a	<u>THYMIDYLATE SYNTHASE (TS)^b</u>		FBC ^c	
	pmol/g	% of Baseline	nmol/g	% of Baseline
0 min	1.31 ±0.13	(100)	5.88 ±0.56	(100)
10 min	0.26 ±0.17	19.8	0.23 ±0.02	3.9
20 min	0.56 ±0.06	42.7	0.27 ±0.01	4.6
40 min	0.99 ±0.08	75.6	0.21	3.6
60 min	1.47 ±0.13	112.2	0.14 ±0.01	2.3

Biopsies of solitary skin metastasis, average weight 68 ± 58 mg, time after CH₂FH₂ administration.

• By [6^{-3}H]FdUMP ligand-binding assay (CP Spears et al., Cancer Res. 42:450-56 (1982)).

Folate Binding Capacity, FBC, is a measure of tissue CH₂FH₄ and FH₄ level (Invest. New Drugs 7:27-36 (1989), (modified after Priest et al., Biochem. J. 216:295-98 (1983))), with a Sigma (6R,S)-CH₂FH₄ standard value of 936 DPM/pmole.

TABLE III

TS INHIBITION AFTER CH₂FH₄ ADMINISTRATION

PATIENT: K.H.; last 5-FU treatment: ≥ 1 week
 LOCATION: Östra Sjukhuset (Eastern Hospital), Sweden
 TUMOR: Rectal adenocarcinoma, locally advanced
 CH₂FH₄ FORMULATION: 0.2 M Na Ascorbate, Sigma (6R,S)-CH₂FH₄
 CH₂FH₄ DOSE: 35 mg IV over 1 min week #1; 50 mg IV in
 40 ml week #2

(Tumor Tissue Values = Ave. ± S.D.)

Time of Biopsy ^a	THYMIDYLATE SYNTHASE (TS) ^c		Δ DPM Week #1	FBC ^b Week #1	% of Baseline Week #2
	pmol/g Week #1	% of Baseline Week #2			
0 min	5.77 (100) ±0.09	5.64 (100) ±1.26	759 (100) ±145	499 (100) ±190	
10 min	6.28 (212.4) ±1.92	10.25 (181.7) ±0.82	320 (42.2) ±60	376 (75.4) ±17	
20 min	2.26 (43.7) ±0.36	5.91 (104.8) ±0.17	314 (41.4) ±9	814 (163.1)	
30 min	5.90 (114.1) ±0.12	2.02 (35.8) ±0.03	632 (83.3) ±26	249 (49.9) ±75	
40 min		3.46 (61.3) ±0.28		399 (80.0) ±44	
24 hr	6.32 (122.2) ±0.52		1403 (184.8) ±130		

- On Week #1 the CH₂FH₄ was formulated at pH 2.0, DEAE-purified; On Week #2 the preparation was pH 9.0, with 6 mM (final concentration) CH₂O added, no DEAE step used.
- Biopsies of rectal pouch mass, average weights, 145 ± 39 mg (Week #1) and 136 ± 24 mg (Week #2). Time after CH₂FH₄ administration.
- By [6-³H]FdUMP ligand-binding assay (Spears et al., Cancer Res. 42:450-56 (1982)).
- Folate Binding Capacity, given in ΔDPM over [³H]FdUMP-TS binary complex background (Invest. New Drugs 7:27-36 (1989)); standard curve Sigma (6R,S)-CH₂FH₄ showed 920 and 898 ΔDPM/pmole for weeks 1 and 2. Multiply ΔDPM values by 0.0002 to convert to nmol/g.

In patient A.M., a sixty-seven year old woman with over a 3 year prior history of disseminated gastric cancer, and who was end-stage in her course, TS was inhibited 80.1 and 57.3 % in 5 her tumor at 10 and 20 min, respectively, in her tumor after CH₂FH₄ administration. (It should be noted that the CH₂FH₄ preparation was over 85% FH₄.) Notably, when she was studied again 2 weeks subsequently, with a repeat dose of CH₂FH₄, TS in the 10 baseline tumor biopsy was undetectable (data not shown).

The FBC (folate binding capacity of L. casei TS-[3H]FdUMP added to the cytosols, (a measure of tissue CH₂FH₄ and FH₄, mostly presumed to 15 be polyglutamates) also showed a surprising decrease, which continued through 60 min. Tissue FH₄ polyglutamates were not separately measured by use of CH₂O addition to the FBC conditions. The continuing drop in FBC, however, at the 60-min time 20 point rules out the possibility that all post-CH₂FH₄ biopsies were somehow an artifact of tumor tissue sampling. This paradoxical decrease in FBC is a characteristic feature of 5-FU-responding patients receiving high-dose LV added to 5-FU bolus i.v. 25 therapy (C.P. Spears, et al. Presentation at 25th Annual Am. Soc. Clin. Oncol. meeting, May 22, 1989). This decrease was also seen in tumor of patient K.H. (Table 3). An explanation for the paradoxical decrease in FBC is that one-carbon exchange (e.g., 30 R.G. Matthews et al, Adv. Enz. Regul. **26**:157-70 (1987) occurred in the tumor tissue, between FH₄-monoglutamate derived within minutes from administration of the CH₂FH₄/FH₄ drug, and endogenous CH₂FH₄-polyglutamates. Since the polyglutamates of 35 CH₂FH₄ may be expected to bind TS-FdUMP up to 50-fold more strongly than the monoglutamate (Houghton et al., Cancer Res. **48**:3062-69 (1988)), the one-

carbon exchange could lead to the observed decrease. This data is powerful evidence that CH₂FH₄/FH₄ given to this patient was rapidly transported and metabolized in her tumor. The decrease in TS in her tumor, then, is assumed to be related to this metabolism and the presence of non-measurable levels of FdUMP (at concentrations near stoichiometry with endogenous TS binding sites). The paradox of decreasing free TS with decreasing FBC also can be explained by metabolic channeling of administered CH₂FH₄ (Reddy et al., Proc. Natl. Acad. Sci. USA 77:3312-16, 1980), or by formation of TS-FdUMP-tetrahydrofolate, or of TS-deoxyuridylate-CH₂FH₄, ternary complexes by the unnatural (6S)-CH₂FH₄ or (6R)-FH₄ enantiomer, or by TS-FdUMP-CH₂FH₄ due to very rapid ternary complex formation (Lockshin et al., Biochem. Pharmacol. 30:247-57 (1981)) prior to the 10-min biopsy sample and one-carbon folate metabolism. In fact, the last explanation may be the most attractive, since the maximum TS inhibition was at this first biopsy time point. The degree of TS inhibition, 80.2% decrease over baseline value, and relatively limited duration of TS inhibition would predict that higher concentrations of FdUMP (as would result from 5-FU given shortly before, or with the CH₂FH₄) would lead to the desired therapeutic objective of complete TS inhibition.

In patient K.H., a fifty-five year old man with locally unresectable advanced rectal adenocarcinoma, the TS pharmacodynamic tumor tissue analyses were done twice, nine days apart. Following study, K.H. continued to receive intermittent bolus 5-FU. This patient had been previously a partial responder to 5-FU plus LV, with stable disease at the time of initial CH₂FH₄ administration. There were modifications of the CH₂FH₄ formulation between the 2 pharmacodynamic

studies (See Table III). In the first study week, the pH was not adjusted up from 2.0, after DEAE column isolation of the Sigma (6R,S)-CH₂FH₄. Thus, some of this folate may also have been 5,10-methenyl-tetrahydrofolate. In the second study week, the pH was adjusted up to 9.0, and no DEAE step was used (with therefore 6 mM formaldehyde being present in the 40-cc volume for injection).

Patient K.H. showed changes in TS and in FBC assays after CH₂FH₄ administration that were qualitatively similar to those of Patient A.M., shown in Table III. Again, significant inhibition of TS over baseline values occurred in tumor samples after the CH₂FH₄ was given, in the absence of recent 5-FU exposure. On the first occasion, however, the pH of the formulation was low, and possibly the CH₂FH₄ was less well solubilized (or less stable, or both) than on Week #2, when an alkaline pH was used in addition to an excess of CH₃O. Comparison with patient A.M. suggests that the acute TS decrease resulted from FH₄ rather than CH₂FH₄. As in Patient A.M., TS inhibition, on both occasions, was transient, averaging 36 to 44% of baseline values for the combined data of the two studies, during the 20 to 30 min period after CH₂FH₄ was given. The most significant evidence of an increase in CH₂FH₄, as reflected by FBC assay, was at 24 hr after the first dose, which was expected on the basis of slow polyglutamation of folates generally. Significant drops in FBC also occurred in both weeks of study, again suggestive of the postulated one-carbon exchange between drug-monoglutamates and endogenous CH₂FH₄-polyglutamates. The fact of a less striking change in FBC values in tumor biopsies from K.H. than in A.M. is also consistent with the lower baseline FBC values (given in raw DPM, multiply by 0.0002 to convert to nmol/g units comparable to

Patient A.M.), and the less striking but highly significant TS inhibition in tumor of K.H. As with Patient A.M., the data would predict, using purely kinetic arguments, that higher FdUMP levels

5 generated from 5-FU given closer to the time of CH₂FH₂ dosing would lead to desired abrogation of TS activity.

It has long been known that FdUMP tends to persist at low levels in tissues following a single
10 dose of 5-FU. FdUMP may therefore be slowly released from the RNA storage compartment inside cells.

Thus, because only trace concentrations of FdUMP are required to inhibit TS, if CH₂FH₂ or FH₂ levels are high, the TS inhibition observed in these 15 two patients was likely to have been due to facilitation by the natural (6R)-CH₂FH₂ or (6S)-FH₂ enantiomers (diastereomers) of the CH₂FH₂ formulation on TS binding by residual FdUMP levels. These 20 results suggest that repeated administration of CH₂FH₂ or FH₂ may be as effective as repeated dosing with 5-FU, but without the toxicity of dose-escalation of 5-FU.

The patients who received CH₂FH₂ showed no
25 acute toxicities due to this treatment, including the instance of week #2 in K.H. when a slight excess of CH₂O was present in the preparation. However, they did continue to manifest the same toxicities as their prior experience with 5-FU plus LV (i.e., mild
30 nausea and fatigue). Patient A.M., as noted above, had extremely advanced gastric cancer at the time of the study and so was not evaluable for response. However, patient K.H. showed endoscopic evidence of continued disease stabilization if not at least
35 additional, minor tumor regression noted over the subsequent months after the two weeks of CH₂FH₂ administration.

34

Example 4(6R,S)-FH, ADMINISTRATION TO RATS BEARING
TRANSPLANTED HEPATIC COLONIC CARCINOMAS

Table IV (below) shows the results of (6R,S)-FH, (see Figure 3) administration to rats bearing transplanted hepatic colonic carcinoma. The present inventors have considerable experience with this model, and the antitumor effects of 5-FU shown are typical results, as are the TS and folate assays of control and 5-FU-only-treated rats. A striking finding was of growth stimulation yet decreased TS levels after (6R,S)-FH, alone. In fact, the "free TS" levels in the (6R,S)-FH,-only-treated rats were the lowest of all arms of the study. This observation suggests that either the natural 6S-FH, or the unnatural 6R-FH, may have formed TS-inhibitory TS-dUMP-folate ternary complexes. In combination, the degree of synergy of (6R,S)-FH, with 5-FU in this example appears to be greater than previously found for (6R,S)-leucovorin (Carlsson et al., Anticancer Res. 10:813-16 (1990)).

TABLE IV

(6R,S)-TETRAHYDROFOLATE^a AS A MODULATOR OF 5-FU
IN AN EXPERIMENTAL LIVER CANCER IN RATS^bRESULTS AT DAY 17 AFTER TRANSPLANTATION
(Average of 3 rats/treatment)

	<u>TREATMENT</u>	<u>TUMOR WEIGHT</u>	<u>TS^c</u>	<u>5,10-CH₂FH^d</u>	<u>FH^e</u>
		(g)	(p mole/g)	(nmol/g)	(nmol/g)
30	CONTROL	5.84	18.96	0.69	1.18
	5-FU ONLY (30 MG/KG)	1.03	9.03	4.11	2.39
	5-FU ^c + (6R,S)-FH ^c	0.31	9.23	1.23	1.76
35	(6R,S)-FH only (30 mg/kg)	10.43	7.13	2.93	2.31

35

5 ^a (6R,S)-FH, was the commercially available racemic tetrahydrofolate from Fluka Chemical Corp. (Cat. No. 87355, "Tetrahydrofolic acid dihydrochloride monohydrate," or "5,6,7,8-Tetrahydropteroyl-L-glutamic acid dihydrochloride monohydrate," >94% by HPLC). The (6 R,S)-FH, was weighed, dissolved in normal saline, and injected Days 2-5 by tail vein administration using the air-free Protector device 10 to prevent oxidative destruction of the folate.

15 ^b Inoculation of 1×10^6 viable colon tumor (nitrosoguanidine-induced) cells under the liver capsule on Day 1 (Carlsson et al., Anticancer Res. 10:813-16 (1990)). Animals sacrificed on Day 17 for excision of single liver tumor nodules for pharmacodynamic studies.

20 ^c 30 mg/kg
^d Assays done as described (Spears et al. Adv. Exp. Med. Biol. 244:98-104 (1988)) and done at 24 h after injection.

Example 5

Spontaneous Conversion of CH₂FH₂ to FH₂ by Dilution

25 Figure 4 shows the results of TS-[³H]FdUMP-folate binding assay of CH₂FH₂ as a function of concentration of the folate in 0.2 M Tris buffer, pH 7.4, with and without formaldehyde (CH₂O), 6 mM, addition. The CH₂FH₂ was prepared as the racemic (6R,S) material from (6R,S)-FH₂ and excess formaldehyde, and DEAE-column isolation as described 30 in Figure 1. This preparation was essentially free of free formaldehyde based on colorimetric assay of bulk material (Nash, Biochem. J. 55:416-21 (1953)).

35 At all concentrations (total assays volume 150 μ l), excess formaldehyde was required to obtain maximal binding (which was still only 19.3% of stoichiometric binding). A notable effect was the increasing need for formaldehyde addition with increasing dilution, to obtain maximal CH₂FH₂ assay recovery.

36

This phenomenon has been a repeated observation in the laboratories of the inventors, and clearly shows that CH₂FH, on dilution becomes FH, with liberation of free formaldehyde. The concentration requirement for formaldehyde to reverse the FH, formation caused by dilution is in the millimolar range which is vastly higher than physiologic.

This requirement for a large excess of formaldehyde to shift the equilibrium between FH, and CH₂FH, (Eq. 1) was found by the inventors to



be independent of temperature, pH or formaldehyde content of charcoal isolation, the presence of air exposure, or the presence of reducing agents. In addition, [11-¹⁴C]CH₂FH, prepared as described (Moran et al., Proc. Natl. Acad. Sci. USA 76:1456-60 (1979)), and DEAE-purified (as the concentrated material) of excess "CH₂O, was confirmed to have a labile 14CH₂O group by dimedone trapping. For instance, 46,664 DPM of [11-¹⁴C]-CH₂FH, diluted to 1 ml in H₂O was found to have 67.8% of the label recoverable by chloroform extraction of dimedone (methone) product (37°C).

CLAIMS:

1. A method of inhibiting the growth of a tumor in a patient comprising administering to said patient an amount of 5,10-methylene-tetrahydrofolate (CH₂FH₄) and 5-Fluorouracil (5-FU) sufficient to effect said growth inhibition.

5 2. The method of claim 1 wherein CH₂FH₄ is administered to said patient concurrently with 5-FU.

10 3. The method of claim 1 wherein CH₂FH₄ is administered to said patient prior to the administration of 5-FU.

4. The method of claim 3. wherein CH₂FH₄ is administered to said patient 6-24 hours prior to the administration of 5-FU.

15 5. The method of claim 4 wherein CH₂FH₄ is administered to said patient 1-3 hours prior to the administration of 5-FU.

20 6. The method of claim 1 wherein CH₂FH₄ is administered to said patient subsequent to the administration of 5-FU.

7. The method of claim 6 wherein CH₂FH₂ is administered to said patient 1-10 days subsequent to the administration of 5-FU.

8. The method of claim 7 wherein CH₂FH₂ is 5 administered to said patient 1-6 hours subsequent to the administration of 5-FU.

9. The method of claim 1 wherein CH₂FH₂ is administered to said patient intravenously, intraarterially or intraperitoneally.

10 10. The method of claim 9 wherein CH₂FH₂ is administered in a dosage of 5-500 mg/m².

11. The method of claim 10 wherein CH₂FH₂ is administered in a dosage of 20-200 mg/m².

15 12. The method of claim 10 wherein CH₂FH₂ is administered intravenously.

13. The method of claim 12 wherein CH₂FH₂ is administered to said patient every 4-6 hours.

14. The method of claim 12 wherein CH₂FH₂ is administered to said patient once daily.

15. The method of claim 12 wherein CH₂FH,
is administered to said patient once weekly.

16. The method of claim 13 wherein CH₂FH,
is administered prior to the administration of 5-FU.

5 17. The method of claim 13 wherein CH₂FH,
is administered subsequent to the administration of
5-FU.

10 18. The method of claim 14 wherein CH₂FH,
is administered to said patient through a central
venous catheter.

19. The method of claim 1 wherein CH₂FH,
is administered to said patient as the 6R
diastereomer, the 6S diastereomer, or a mixture of
the 6R and 6S diastereomers.

15 20. The method of reducing toxicity of an
anti-folate drug in a patient administered said drug
comprising administering to said patient an amount
of CH₂FH, sufficient to reduce said toxicity.

20 21. The method of claim 20 wherein the
anti-folate drug is methotrexate, trimetrexate,
nitrous oxide or dideoxytetrahydrofolic acid.

40

22. A method of treating folate deficiency comprising administering to a patient in need of such treatment an amount of CH₂FH₂ sufficient to effect said treatment.

5 23. The method of claim 1 wherein the concentration of CH₂FH₂ administered is from 0.1 to 20 mg/ml in alkaline vehicles.

10 24. The method of claim 1 wherein the concentration of CH₂FH₂ administered is from 0.1 to 10 mg/ml in physiologic saline.

25. A method of treating B12- and B6-refractory anemias comprising administering to a patient in need of such treatment an amount of CH₂FH₂ sufficient to effect said treatment.

15 26. A composition comprising an amount of CH₂FH₂ and 5-FU sufficient to inhibit tumor growth in a patient together with a pharmaceutically active carrier.

20 27. The composition of claim 26 further comprising an agent that stabilizes CH₂FH₂.

28. The composition of claim 27 wherein
the agent that stabilizes CH₂FH₂ is an ascorbate
salt.

29. The composition of claim 27 wherein
5 the agent that stabilizes CH₂FH₂ is reduced
glutathione.

30. The composition of claim 26 further
comprising formaldehyde.

31. A composition comprising an amount of
10 CH₂FH₂ and a drug which is metabolized to
fluorodeoxyuridylate (FdUMP) sufficient to inhibit
tumor growth in a patient together with a
pharmaceutically active carrier.

32. The composition of claim 31 wherein
15 the drug which is metabolized to FdUMP is
floxuridine (FUDR), fluorafur, or 5'-
deoxyfluorouridine.

33. The method of claim 9 wherein CH₂FH₂
is administered to said patient by protracted,
20 continuous venous infusion through a central venous
catheter.

34. A method of inhibiting the growth of
a tumor in a patient comprising administering to
said patient an amount of tetrahydrofolate (FH₄) and
5-Fluorouracil (5-FU) sufficient to effect said
growth inhibition.

35. The method of claim 34 wherein FH₄ is
administered to said patient concurrently with 5-FU.

36. The method of claim 34 wherein FH₄ is
administered to said patient prior to the
administration of 5-FU.

37. The method of claim 36 wherein FH₄ is
administered to said patient 6-24 hours prior to the
administration of 5-FU.

38. The method of claim 37 wherein FH₄ is
administered to said patient 1-3 hours prior to the
administration of 5-FU.

39. The method of claim 34 wherein FH₄ is
administered to said patient subsequent to the
administration of 5-FU.

40. The method of claim 39 wherein FH₄ is
administered to said patient 1-10 days subsequent to
the administration of 5-FU.

43

41. The method of claim 40 wherein FH_x is administered to said patient 1-6 hours subsequent to the administration of 5-FU.

5 42. The method of claim 34 wherein FH_x is administered to said patient intravenously, intraarterially or intraperitoneally.

43. The method of claim 42 wherein FH_x is administered in a dosage of 5-500 mg/m².

10 44. The method of claim 43 wherein FH_x is administered in a dosage of 20-200 mg/m².

45. The method of claim 43 wherein FH_x is administered intravenously.

46. The method of claim 45 wherein FH_x is administered to said patient every 4-6 hours.

15 47. The method of claim 45 wherein FH_x is administered to said patient once daily.

48. The method of claim 45 wherein FH_x is administered to said patient once weekly.

49. The method of claim 46 wherein FH, is administered prior to the administration of 5-FU.

50. The method of claim 46 wherein FH, is administered subsequent to the administration of 5-FU.

51. The method of claim 47 wherein FH, is administered to said patient through a central venous catheter.

52. The method of claim 34 wherein FH, is administered to said patient as the unnatural 6R diastereomer, the natural 6S diastereomer, or a mixture of the 6R and 6S diastereomers.

53. The method of claim 34 wherein the concentration of FH, administered is from 0.1 to 20 mg/ml in alkaline vehicles.

54. The method of claim 34 wherein the concentration of FH, administered is from 0.1 to 10 mg/ml in physiologic saline.

55. A composition comprising an amount of FH, and 5-FU sufficient to inhibit tumor growth in a patient together with a pharmaceutically active carrier.

45

56. The composition of claim 55 further comprising an agent that stabilizes FH₄.

57. The composition of claim 56 wherein the agent that stabilizes FH₄ is an ascorbate salt.

5 58. The composition of claim 56 wherein the agent that stabilizes FH₄ is reduced glutathione.

59. The composition of claim 56 wherein the agent that stabilizes FH₄ is formaldehyde.

10 60. A composition comprising an amount of FH₄ and a drug which is metabolized to fluorodeoxyuridylate (FdUMP) sufficient to inhibit tumor growth in a patient together with a pharmaceutically active carrier.

15 61. The composition of claim 60 wherein the drug which is metabolized to FdUMP is floxuridine (FUDR), florafur, or 5'-deoxyfluorouridine.

20 62. The method of claim 34 wherein FH₄ is administered to said patient by protracted, continuous venous infusion through a central venous catheter.

- 1 / 4 -

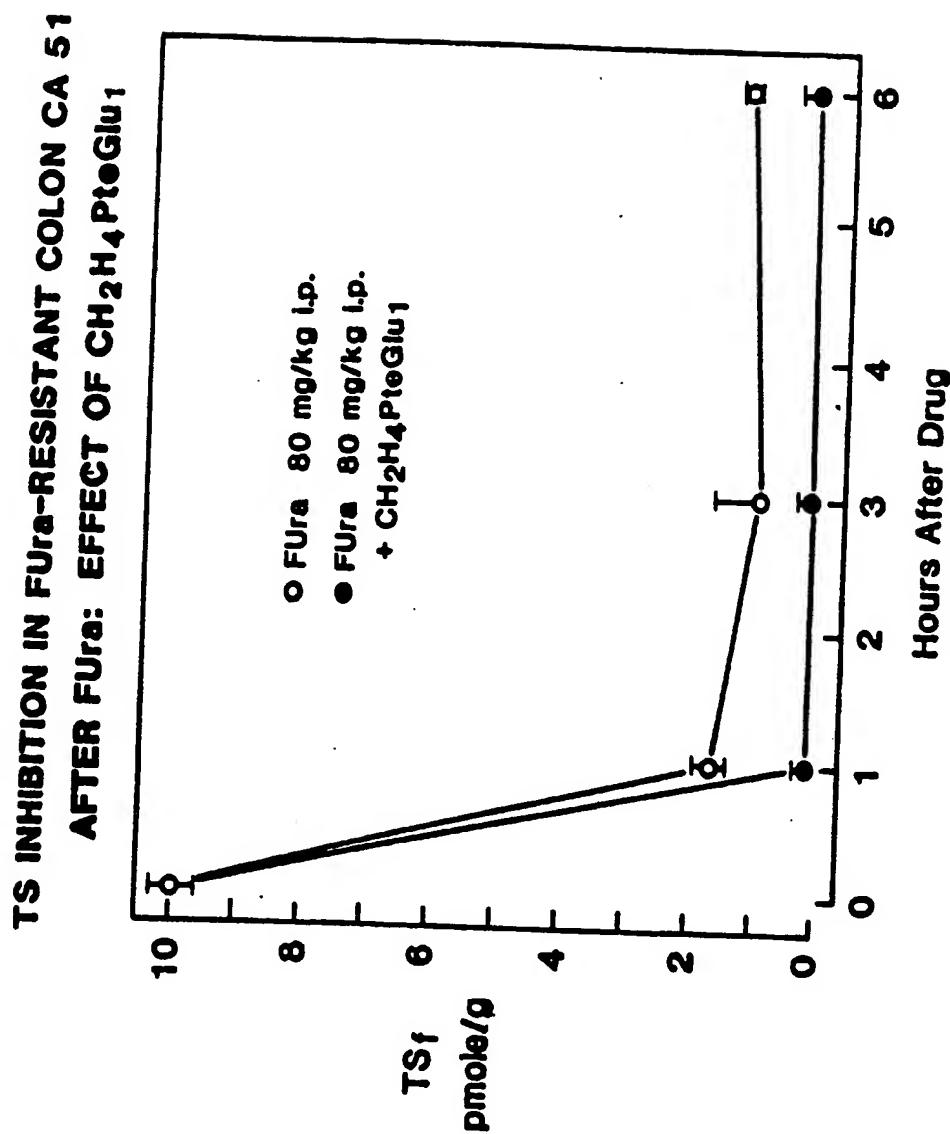
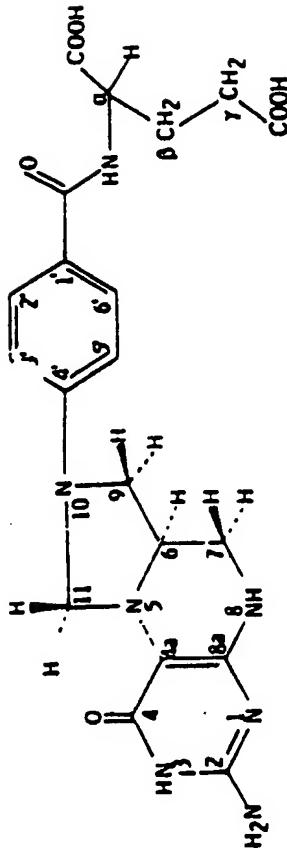
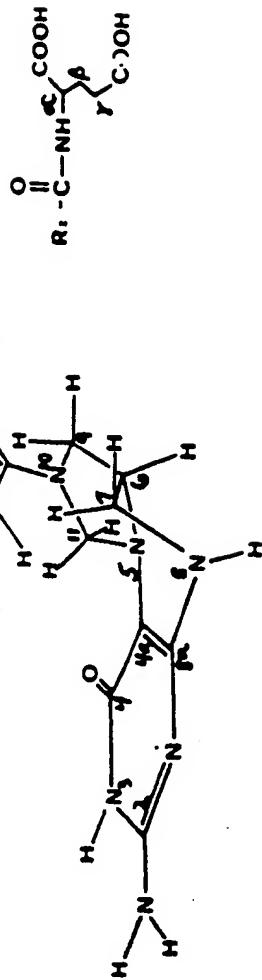


Fig. 1

- 2 / 4 -



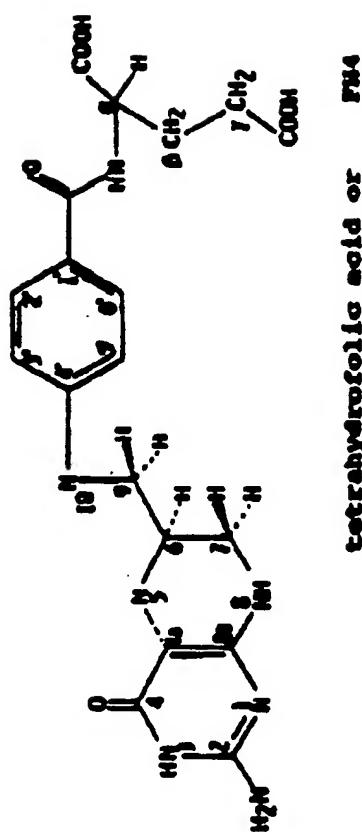
(6R, S)-methylene-tetrahydrofolic acid or CH₂FH₄



Configuration of the natural (6R)-CH₂FH₄ enantiomer

Fig 2

- 3 / 4 -



- 4 / 4 -

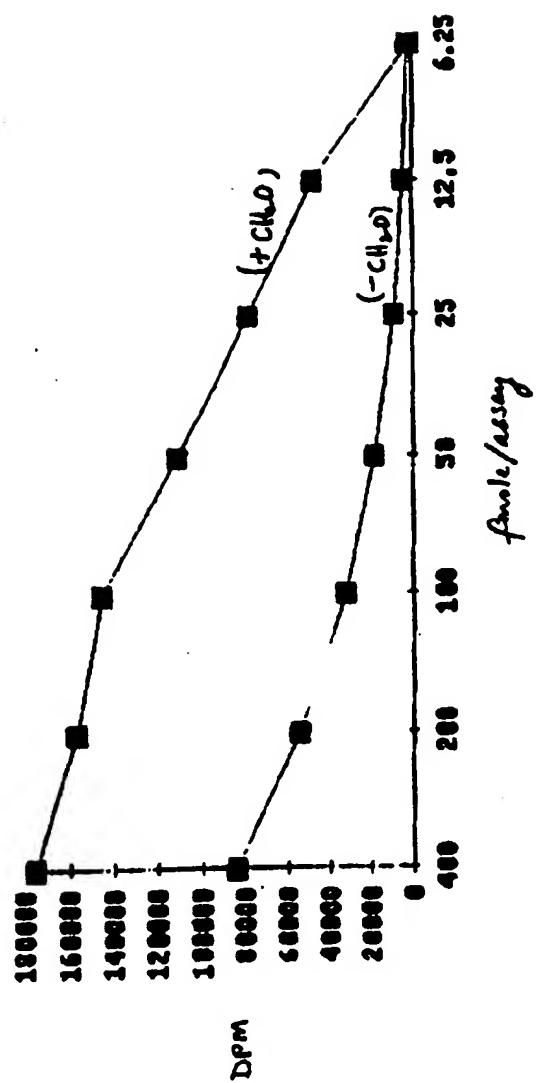


Fig. 4

INTERNATIONAL SEARCH REPORT

International Application No. PCT/US91/03186

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) ⁸

According to International Patent Classification (IPC) or to both National Classification and IPC

IPC(5): A01N 431/54

US : 514/274

II. FIELDS SEARCHED

Minimum Documentation Searched ⁷

Classification System	Classification Symbols
U.S.	514/274

Documentation Searched other than Minimum Documentation
to the Extent that such Documents are Included in the Fields Searched ⁸

CAS ON LINE: A.P.S.

III. DOCUMENTS CONSIDERED TO BE RELEVANT ⁹

Category ¹⁰	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³
X, Y	Spears, et al, <u>Method for Thymidylate-Synthase Pharmacodynamics: Serial Biopsy, Free and total T,S Fdump, and H₄ PTEGLUM and CH₂-H₄ PTEGLU Assays</u> ; Adv. Exp. Med. Biol.; 244:98-104(1988) See entire document	1-62
Y	Spears, et al <u>Activation of Leucovorin (CF) To Methylenetetra hydrofololate (CH₂FH₂) for Improving Thymidylate Synthase (TS) Inhibition after 5-F4: Effects of CF Dose, L-Serine, L-Glutamate, and direct Methyl-Tetrahydro-folate (CH₂FH₂) Administration</u> , Proceedings of Asco, Vol. 8, March 1989 (#269), pg. 69. See entire document	1-62
Y	Grem et al, <u>Overview of Current Status and Future Director of Clinical Trials with 5-Fluorouracil in combination with Folinic Acid</u> , Cancer Treatment Reports, Vol 71, No. 12, December, 1987 Pgs. 1249-1264. See entire document	
Y	Machorer, et al, <u>Treatment of Advanced Colorectal and Gastric Adrenocarcinomas with 5-FM combined with High-Dose Folinic Acid: A Pilot study</u> , Cancer Treatment (Cont. on second sheet)	1-62

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- "&" document member of the same patent family

IV. CERTIFICATION

Date of the Actual Completion of the International Search

13 August 1991

Date of Mailing of this International Search Report

26 AUG 1991

International Searching Authority

ISA/US

Signature of Authorized Officer ¹⁴

Theodore J. Criares

III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET)

Category *	Citation of Document, with indication, where appropriate, of the relevant passages	Relevant to Claim No
	(Cont. from 2nd sheet) Reports, Vol. 66, No 10, October, 1982, Pgs 1803-1807. See entire document	

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